Journal of Chromatography, 146 (1978) 439-448 Biomedical Applications Q **.Blsevier Scientific Publishing Gompauy, Amsterdam - Rinted in The Netherlands**

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RAPID AND SENSITIVE DETERMINATION OF ENZYMATIC DEGRADATION PRODUCTS OF ISOMERIC CHONDROmIN SULFATES BY HIGH-PERFORMANCE LIQUID CHROMATOGRAPHY

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(Received May 8th, 1978)

SUMMARY

The separation and quantitative analysis of enzymatic degradation products of isomeric ehoadroitin sulfktes by high-performance liquid chromatography (HFLC) are described. The substituted unsaturated disaccharides which result from digestion of chondroitin sulfates with chondroitinase are quickly separated on polar adsorbents such as silica gel. The UV absorption properties of these uusatunted disaccharides permit *W* **measurement with** detection limits of approximately 100 ng. Their separation by HPLC facilitates the use of **enzymatic methods for the determination of chondroitin sulfates A, B and C.**

The potential of this method in clinical application is demonstrated by quantitative assays of glycosaminoglycans from a normal urine and urine from a patient with Hunter syn**drome. The results are consistent with amount of isomeric chondroitm sulfates found in comparable urines by others.**

INTRODUCTION

Ghondroitin sulfates A, B and C'* are isomeric mucopolysaccharides containing alternating hexuronic acid and N-acetylgalactosamine residues as the

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The abbreviations used are: ChS-A, ChS-B, and ChS-C = chondroitin sulfates A (chondroitin 4-sulfate), B (dermatan sulfate), and C (chondroitin 6-sulfate); GAG = glycosamino $glycan(s)$ [mucopolysaccharide(s)]; $CPC = cetylpyridinium chloride; \Delta Di-4S = 2-acctamido-$ **2-d~xy-3-0-(8-~gluco-44nepyranosylumnic acid)+G-sulfo-D-galactose; ADi-6S = 2 acetamido-2-deoxy-3-O-(6-D-gluco-4-enepyranosyluronic acid)-6-O-sulfo-D-galactose; ADi-OS **= 2-acetamido-2-deo~-~-O~~-D~uco-49nepyranosgluohic acid)-D-galactose; ADi_diS,** $= \Delta 4.5$ -sulfoglucuronido-acetylgalactosamine-4-sulfate.

charact&i&ic structural batikbone. The suifate residue is located at C-4 of the galactosamine residue in ChS-A and B and at C-6 in ChS-C $[1, 2]$. The hexur**onic acid residue is derived from gIucuronic acid in ChSGA and .C and from** both iduronic acid and glucuronic acid in ChS-B $13-51$. Because of the sim**ilarity** in structure of the three isomeric chondroitin sulfates, few convenient methods have been reported for the quantitative analysis of the individual isomers in mixtures $6-91$. An enzymatic method for the determination **of these chondroitin &fates was developed by Saito et al. [lo], who purified chondroitinase ABC from Proteus** *vulgaris* **and chondroitinase AC from Flavo***bacterium heparinum* **which both specifically degraded chondroitin &fates into the unsaturated disaccharides [ll] .**

The same $4,5$ -unsaturated disaccharide 4 -sulfate $(\Delta \text{Di-4S})$ is formed from **both ChS-A and ChS-B. The 4.5-unsaturated disaccharide 6-sulfate (ΔDi-6S) is produced only from ChS-C [lo]** . **The assay of the unsaturated disaccharides in the enzymatic degradation product mixture by anaiytical procedures leads** to the identification and measurement of glycosaminoglycans. This enzymatic **method has been widely used to estimate giycosaminoglycans in tissues and** in serum, urine and other body fluids $[12-16]$.

Methods for the determination of the unsaturated disaccharides from the enzymatic degradation mixture have usuaIIy employed paper chromatography [lo, 17, 181, and have required up to 36 h for the development of the chromatograms. The detection Iimit is approximately 25 pg. Recently, thin-layer chromatography has been employed for these assays [19]. Although the analysis was improved by shortening the chromatograph developing time and increasing detection sensitivity, the method required 12 h including a desalting step. These procedures did not provide detailed quantitative data. Calorimetric methods for the assays of sulfated disaccharides in the enzymatic digest by forming chromogens with both sulfated disaccharides and **with** *only* **ADi-GS were also reported recentiy [20]** . This **method appears to be sensitive but does not effectively differentiate between the individual disaccharides. The HPLC method described here is rapid and sensitive and can be used to estimate quantities of each of the different chondroitin sulfates.**

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Materials

Chondroitin sulfates A and B and the unsaturated disaccharides were pur**chased from Miles Labs. (RJkhsrt, Ind,, U_SA_). Chondroitin &fate C was purchased from Calbiochem (San Diego, C&f., U.S.A.). Chondroitinase ABC** and AC were obtained from both Miles Labs. and Sigma (St. Louis, Mo., U.S.A.).

Prepacked HPLC columns, Partisil PXS, $10 \mu m$, $25 \text{ cm} \times 4.6 \text{ mm}$ I.D. (Whatman Labs., Clifton, N.J., U.S.A.) and SI-5A, $5 \mu m$, $25 \text{ cm} \times 4.6 \text{ mm}$ I.D. (Brownlee Labs., Berkeley, Calif., U.S.A.) were employed. All solvents used were distilled in glass (Burdick & Jackson Labs., Muskegon, Mich., U.S.A.)

Instruments

For HPLC determination, a Waters Assoc. Model 6000A liquid chromato-

Isolation of urinary glycosaminoglycans

The. *urine* **specimens were filtered through** *Whatman No.* **2 filter paper and adjusted to pH 5.0 with 0.5** *M* **acetic acid. Two milliliters of the treated** urine was mixed with $30 \mu l$ of 5% cetylpyridinium chloride *(CPC)* in 0.9% **NaCl. The urinary GAG required 12 h equilibration at 4" with CPC for maximum precipitation as a CPC-polysaccharide complex [21]. After standing overnight at 4° the precipitate which formed was washed three times with 0.1% CPC and dissolved in 1 ml 2.0** *M* **NaCl. The insoluble material was .centrifuged. Four volumes of absolute ethanol were added to the supematant and GAG were precipitated at 0" for 4 h. The precipitate was wsshed successively with 80% ethanol, absolute ethanol and ether, and dried under a stream of nitrogen. The GAG obtained were further dissolved in 0.5 ml distilled** water and precipitated overnight at 0° in four volumes of absolute ethanol **saturated with sodium acetate. The precipitate was washed with absolute** ethanol and ether and dried in vacuum over $P_2 O_5$. The GAG isolated was **then used for enzymatic digestion.**

Enzymatic digestion

The digestion mixture contained 10 μ l of a solution of 100 μ g standard GAG or GAG from 2 ml of urine in water, $10 \mu l$ of enriched Tris buffer (pH **8.0)** [10] and 20 μ of an aqueous solution of either chondroitinase ABC **or chondroitinsse AC (10 units/ml). After incubation at 37" for 2.5 h, another lo-p1 portion of enzyme solution was added and incubation continued for another 2.5 h at 37" [19]. The enzymes were omitted in the blank. Four volumes of absolute ethanol were then added and the mixtures left overnight at 4". The clear supematant obtained by centrifugation was dried under a** stream of nitrogen and the residue, dissolved in $100 \mu l$ of 90% methanol, **was applied on the HPLC instrument.**

KPLC Separation of the disaccharides in the enzymatic digest

Columns packed with Partisil 10 $(10 \mu m)$ silica gel particles) and LiChrosorb SI-100 (5- μ m silica gel particles) were used. The mobile phase consisted **of a three-component mixture: dichloromethane-methanol~monium formate buffer. The enzymatic degradation products in 90% methanol were injected directly onto the column.**

The ultraviolet absorption spectra of the products of chondroitinsse action on ohondroitin *sulfaks as* **a function of pH were studied by Nakada [22]. At a pH 1.8, an absorption maximum at 232 nm was recorded. Because of** relatively strong and broad absorption, the unsaturated disaccharides can be **measured with good sensitivity at 254 nm.**

Separations were csrried out isocraticahy at room temperatures. Details are given separately with each chromatogmm.

RESULTS AND DISCUSSION *.,..*

$Chromatography$

A variety of solvent systems were tested. The best separations were achieved **with the ternary solvent dicblorometbane-metbanol-O.5 M ammonium formate buffer pH 4.8** $(60:34:6, v/v/v)$ **. A separation of the three standard** unsaturated disaccharides \triangle Di-OS, \triangle Di-OS and \triangle Di-4S requires 20 min and **is shown in Fig. 1;** *.I*

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The **capacity ratio (k') values of the three disaccharides with different** content of buffer are shown in Table I. The data demonstrate that a change of 1% in buffer content has a significant effect on peak resolution. It is apparent that the resolution of Δ Di-0S and Δ Di-6S increases and resolution **of ADi-GS and ADi-4S decreases with increasing buffer content,**

Fig. 1. HPLC of the three standard unsaturated disaccharides. $1 = \Delta \text{Di-OS}$; $2 = \Delta \text{Di-OS}$; **3 = ADi-4S; Solvent system: dichlorometbane-methanol-O.5 M ammonium formate (pII 4.8) (60:34:6, v/v/v); Column: LiChrosorb SI-100, 5 pm, 25 cm x 4.6 mm I.D. Flowrate: 2.0 ml/min. Pressure: 1800 p.s.i-; Injection port amount: 5 pg of each disaccharide_** *W* **detection at 254 pm. 0.01** a.u.f.s.

TABLEI

CAPACITY FACTORS MEASURED FOR THE THREE STANDARD UNSATURATED DISACCHARIDES AT DIFFERENT AMOUNTS OF AQUEOUS BUFFER IN THE MO-BILE PHASE

Values given are mean k' values determined from five separate chromatograms. Column: LiChrosorb SI-100, 5 μ m, 25 cm × 4.6 mm LD. Flow-rate: 2.0 ml/min. Pressure: 1800 **ps.i.** *W* **detection at 254 pm, 0.01 au.f.s.** 医上唇肌的 化 ~ 0.1 (Section).

The pH of the ammonium formate buffer also plays an important role in the separation. When the pH was below 3.8, two peaks were observed for each disaccharide which suggests that tautomeric forms were separated. The phenomenon is reversible. At a higher pH of the buffer, only one peak for each disaccharide was observed. Peak shapes become more symmetric as the pH is increased. Above pH 4.0, the characteristic resolution of the three disaccharide peaks is nearly constant.

The dependence of the k' values on methanol content for the three unsaturated disaccharides is shown in Fig. 2. Ternary solvent systems containing less than 30% methanol separated into two phases. With an increase in the methanol content, ADi-6S was shifted toward ADi-0S, which was eluted first and eventually overlapped with it. The k' values decrease with increasing methanol concentration and asymptotically approach a limit.

Quantitation

For quantitative analysis, calibration curves were established at 254 nm for Δ Di-OS, Δ Di-4S and Δ Di-6S. The linearity is excellent over a large concentration range and the plots go through the point of origin. The advantage

Fig. 2. k' Values of the three standard disaccharides as a function of the methanol content of the mobile phase. $1 = \Delta Di-4S$; $2 = \Delta Di-6S$; $3 = \Delta Di-0S$; Solvent system: dichloromethane -1 M ammonium formate (pH 4.8) (55:7) with different amounts of methanol added. Column: Whatman Partisil PXS, 10 um, 25 cm × 4.6 mm I.D. Flow-rate: 2.0 ml/min. Pressure: 800 p.s.i. Injection volume: 15 µl. UV detection at 254 nm, 0.01 a.u.f.s.

of HPLC for quantitative separation is that it is possible to determine products directly from a calibration curve. In paper or thin-layer chromatography, several steps are required after developing the chromatogram to recover and assay the disaccharides [10, 19].

Application to enzyme digest

The potential of the HPLC separations of enzymatic degradation products is demonstrated in Fig. 3. Figs. 3A and 3C illustrate separations of the degradation products obtained in the incubation of ChS-A and ChS-C respectively with chondroitinase ABC. Figs. 3D and 3F illustrate separations after digestion of these with chondroitinase AC. The major peaks correspond to the expected products. Because of structural similarities, it is difficult to achieve a homogeneous preparation of one chondroitin sulfate to give a single disaccharide upon exhaustive digestion with chondroitinase. Thus, degradation of ChS-A

Fig. 3. HPLC of degradation products of chondroitin sulfates with chondroitinase. (A) 15 μ g ChS-A with chondroitinase ABC; (B) 15 μ g ChS-B with chondroitinase ABC; (C) 15 μ g ChS-C with chondroitinase ABC; (D) 10 μ g ChS-A with chondroitinase AC; (E) 10 μ g ChS-B with chondroitinase AC; (F) 10 μ g ChS-C with chondroitinase AC. Peaks: $1 = \Delta Di-OS$; $2 = \Delta Di-6S$; $3 = \Delta Di-4S$; $4 = \Delta Di-diS$ g. Solvent system: dichloromethanemethanol-0.5 M ammonium formate (pH 4.8) (60:34:6). Column: LiChrosorb SI-100, 5 µm, 25 cm × 4.6 mm I.D. Flow-rate: 2.0 ml/min. Pressure: 1800 p.s.i. UV detection at titul. $254 \text{ nm}, 0.01 \text{ a.u.f.s.}$ $\mathcal{O}(\mathcal{E}_\mathbf{z}^{\mathrm{max}})$ $\mathbb{E} \times \mathbb{E}$ $\sim 10^{11}$ nata ta tituli المنابذة المعاونين a shi ne s

with chondroitinase ABC or AC gave some \triangle Di-6S represented by additional **minor peaks, which indicated possibIe contamination with ChS-C. Likewise, contamination with ChS-A in ChS-C preparations was suggested by the chromatogram.**

.Fig. 3B shows that ChS-B gave, as expected, ADi-4S as the major product following incubation with chondroitiase ABC. The last peak. in this chromatogram (peak 4) is probably a disulfated disaccharide $(\Delta Di\text{-}disR)$ reported **to occur in CM-B [12, 231.** It **has a higher** *k'* **than the monosulfated disaccharides as expected because of the polarity of the** *sulfate* **group. Positive identification of this peak is in progress. A chromatogram of the products of di**gestion of ChS-B with chondroitinase AC is shown in Fig. 3E, A small Δ Di-4S **subunit is detected. Since chondroitinase AC is reported to have no activity toward ChS-B [ll] ,** this **ADi-4S subunit could be due to contamination by** ChS-A. A small peak corresponding to Δ Di-6S is observed when a large amount **of enzyme digest (30% of 100 pg substrate) is injected which indicates contamination by a small amount of ChS-C is also possible. Blank runs indicate that disaccharides are not produced by this procedure in the absence of enzymes.**

The recoveries of disaccharides from the enzymatic digest of 100 μ g of **chondroitin s&fates measured by HPLC are given in Table II. At least 70% of ChS-A can be recovered as disaccharides by digestion with chondroitinase ABC and up to 80% of ChS-A is recovered by action of chondroitinase AC-II, an enzyme of chondroitinase AC type.**

TABLEII

RECOVERY OF CHONDROITIN SULFATES AFTER INCUBATION WITH CHONDROITINASE **BY USING BPLC SEPARATION**

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Chondroitin sulfates $(100 \mu g)$ A, B and C were incubated separately with chondroitinase ABC and **AC as described iu Experimental. The resulting products went dissolved in 90% methanol and in** jected onto the HPLC column. Values obtained are the mean of four sets from each digestion product.

***N.&L** 7. Not **measured..** *-.*

An **inzyme of chondroitinase *AC type.* A.

Identification of urinary GAG

The application of HPLC to clinical assay is demonstrated by identification of urinary GAG in normal individuals and in a patient with Hunter syndrome.

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Fig. 4. HPLC of enzymatic degradation products of urinary GAG. (A) GAG from normal urine digestion with chondroitinase ABC; (B) GAG from normal urine digestion with chondroitinase AC; (C) GAG from Hunter's urine digestion with chondroitinase ABC; (D) GAG from Hunter's urine digestion with chondroitinase AC. Peaks: $1 = \Delta Di-0S$; $2 = \Delta Di-6S$; $3 = \Delta Di-4S$; $4 = \Delta Di$ -di S . Solvent system: dichloromethane—methanol—0.5 M ammonium formate (pH 4.8) (60:34:6). Column: LiChrosorb SI-100, 5 μ m, 25 cm × 4.6 mm I.D. Flow-rate: 2.0 ml/min. Pressure: 1800 p.s.i. UV detection at 254 nm, 0.01 a.u.f.s. Injection amounts are equivalent to 1.548 mg creatinine for normal urine and 0.234 mg creatinine for Hunter's urine.

Fig. 4 shows HPLC separations of disaccharides from urinary GAG after digestion with chondroitinase ABC and AC. The Δ Di-4S or Δ Di-6S obtained from normal urinary GAG are formed in the similar amounts in treatments with either enzyme chondroitinase ABC or AC (Figs. 4A and 4B). The results indicate that ChS-A and ChS-C predominate in the urinary GAG of normals and ChS-B is present in smaller amounts. This is consistent with the values obtained by other chromatographic methods. On the other hand, the patient with Hunter syndrome excreted a large amount of ChS-B $[24]$. The ADi-4S from Hunter's urinary GAG by digestion with chondroitinase AC are exclusively liberated from ChS-A (Fig. 4D), whereas ADi-4S obtained by digestion with chondroitinase ABC is liberated from ChS-A and B (Fig. 4C). The large difference in amounts of Δ Di-4S obtained from Hunter's urinary GAG in treatments with each to the two enzymes verifies that the excretion of large amounts of ChS-B is the characteristic of Hunter syndrome. Furthermore,

the appearance of the ADi-diS subunit in digestion of Hunter's urinary GAG with chondroitinase ABC (Fig. 4C) demonstrates the presence of ChS-B in the urine. Heparin sulfate is not detected by our chromatogram because of its resistance to digestion by the chondroitinase enzymes.

Table III shows the amounts of each disaccharide derived from enzymatic digestion of normal and Hunter's urinary GAG compared to excreted creatinine. The ratio of Δ Di-4S and Δ Di-6S from ChS-A and C are approximately 2:1 in normal urine. The urinary excretion of ChS-A and C from the patient with Hunter syndrome is significantly greater. These are essentially in agreement with previously reported enzymatic values [12].

TABLE III

ANALYSIS OF CHONDROITIN SULFATES IN URINE BY HPLC

Glycosaminoglycans were isolated from a normal child and a patient with Hunter syndrome and subjected to enzymatic digestion with chondroitinase ABC and AC as described in Experimental. The amount of disaccharides obtained in the enzymatic digest is given as ug disaccharides per mg of urinary creatinine.

 $*N.M.$ = Not measured.

CONCLUSION

Effective separation and quantitation of the unsaturated disaccharides resulting from digestion of chondroitin sulfates with chondroitinase can be carried out by HPLC. This is a more rapid and sensitive determination than previously reported methods. The strong UV absorption of these disaccharides makes possible their detection at 254 nm without derivatization. The sensitivity can be increased by using a UV monitor at 232 nm.

Since it is possible to identify and distinguish between many of the individual mucopolysaccharidoses by the pattern of urinary GAG excretion we believe the present procedure could be usefully employed in such diag-

noses. Further work on the application of this technique to the quantitative determination of isomeric chondroitin sulfates is in progress.

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ACKNOWLEDGEMENTS

The authors thank Dr. Edwin W. Naylor and Dr. Robert Guthrie for help**ful discussions The authors also acknowledge the participation Of Dr: John Dulaney and Dr_ Robert McCluer in the initial discussions. This work was supported by a grant from the National Institute of Child Health and Human Development, ETD 03967.** and the state of the state of

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